

A. L. Nussbaum, G. Brabazon, T. L. Popper and E. P. Oliveto. Introduction of Oxygen into Ring B of Corticoids.

Page 2723. In footnote (15) read "*Helv. Chim. Acta.*, 38, 381 (1955). In footnote (17) the first sentence should read: "F. B. Colton, in U. S. Patent 2,738,348 (March 13, 1956), prepared an oxide and the corresponding bromohydrin of unspecified configuration from 6-dehydrocortisone acetate."

Page 2724. In col. 2, line 19 from the end, read "5% aqueous sodium sulfite."

Page 2725. In col. 1, line 27, formula should read " $C_{25}H_{29}O_3F$." In col. 2, line 1, for "X" read "IVb." Line 5, read "VII" instead of "VIII" on 17 α ,21-Dihydroxy- $\Delta^{4,6}$ -pregnadiene-3,11,20-trione 21-acetate. Line 32 from the end, preceding α , read semi-colon instead of comma.—A. L. NUSSBAUM.

Robert T. Blickenstaff and Frederic C. Chang. Seroflocculating Steroids. V. Reduction of the Bile Acid Side Chain.

Page 2727. In col. 1, line 8, after word methyl insert "3-cholenate (V) to give." In footnote (12a), line 4, for "7-cholestane" read "7-cholestene."

Page 2730. In col. 1, line 3 from the end, for "XVIIIb" read "XVIIb."—FREDERIC C. CHANG.

Chien-Pen Lo. The Reaction of 5-Ethoxymethylenerhodanines with Amines.

Page 3467. In Table II, compound 11, cols. 1, 2 and 3, for "H H H" read "H H CH₃."—C. P. LO.

Theodore L. Brown. The Infrared Carbonyl Band Intensity in Some Substituted Ethyl Acetates.

Page 3514. Because of an error made in the calculations, the dipole moment of ethyl cyanoacetate was reported incorrectly as 3.04 Debye. The correct value is 3.80 Debye, which corresponds to a calculated dihedral angle of about 90°. None of the conclusions of the paper are altered by this correction.—THEODORE L. BROWN.

Edward Wichers. Report on Atomic Weights for 1956-1957.

Page 4122. In col. 2 of the table, the value for sulfur should be 32.066.

Page 4123. In col. 1 of the Table, the value for carbon should be 12.011 and for manganese 54.94.—EDWARD WICHERS.

George W. Anderson and Rolf Paul. N,N'-Carbonyldimidazole, a New Reagent for Peptide Synthesis.

Page 4423. In line 15 below the formulas, for "0.10" read "0.010."—GEORGE W. ANDERSON.

Vernon F. Raaen and Clair J. Collins. The Carbon Isotope Effect in the Pinacol-Pinacolone Rearrangement: A Reinvestigation.

Page 4433. In col. 1, last text line, for "less" read "greater."—CLAIR J. COLLINS.

F. Marshall Beringer and Marvin Mausner. Diaryliodonium Salts. VIII. Decomposition of Substituted Diphenyliodonium Halides in Inert Solvents.

Page 4536. In Table IV in column headed T, °C., for "80" read "60." Also, in first line of last paragraph, for "80°" read "60°."—F. MARSHALL BERINGER.

Cheves Walling, F. W. Stacey, Saunders E. Jamison and Earl S. Huyser. (1) The Reaction of Olefins with Oxygen and Phosphorus. (2) Chemical Properties of the Reaction Product of Cyclohexene with Phosphorus and Oxygen.

Pages 4543 and 4546. In these two papers the name of the second author has been printed wrongly as FRANK R. STACEY.

C. G. Overberger and Harold Gainer. Azo Compounds. XXII. Synthesis of Neophyl-type Azo Compounds.

Page 4556. In the title, for "XXII" read "XXIII." In footnote (1), for "22nd" read "23rd."—C. G. OVERBERGER.

C. G. Overberger and Harold Gainer. Neophyl-type Azo Compounds; their Decomposition and Rearrangement of the Neophyl-type Radical.

Page 4561. In footnote (1), for "23rd" read "24th."—C. G. OVERBERGER.

Don L. Bunker and Norman Davidson. On the Interpretation of Halogen Atom Recombination Rates.

Page 5091. In col. 2, line below eq. (6), for " k_4 " read " K_4 ."

Page 5092. In col. 2, line 1 of eq. (14), for " $m - 1$ " read " $m = 1$."

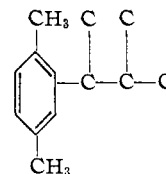
Page 5094. In col. 2, line 23 from the end, for "(1)" read "1."

C. E. Melton and Gus A. Ropp. Studies Involving Isotopically Labelled Formic Acid and its Derivatives. IV. Detection of Cyanide Ion (CN⁻) in a Mixture of Formic Acid and Nitrogen in the Mass Spectrometer.

Page 5573. In col. 1, text line 14, for "production" read "product ion."

B. S. Friedman, F. L. Morritz, C. J. Morrissey and R. Koncos. The Alkylation of *p*-Xylene.

Page 5868. In col. 1, formula II should be



Page 5870. In col. 1, line 3, for "0.2 ml." read "0.2 mole." Five lines below the formulas, omit the duplicating 3-line paragraph.—B. S. FRIEDMAN.